

Q1.

| Question number | Answer | Additional guidance | Mark |
|-----------------|---|------------------------------------|------|
| | The time taken for the activity of a radioactive nuclide to halve (1) | accept for nuclide: isotope sample | (1) |

Q2.

| | Answer | Acceptable answers | Mark |
|--|--------|--------------------|------|
| | A | | (1) |

Q3.

| | Answer | Additional guidance | Mark |
|-----|---------|---------------------|------------|
| (i) | 260 (g) | | (1) AO2 |

| | Answer | Additional guidance | Mark |
|------|---|--|----------------|
| (ii) | (54 days is) 3 half-lives (1) 65 (1) | 260 ÷ 2 (÷ 2) or 520 ÷ 2 ÷ 2 (÷ 2) 18, 36, 54 (represents 3 half-lives) 54/18 = 3 (half-lives) ecf answer to 4ci ÷ 4 130 scores 1 mark award full marks for the correct answer without working | (2) AO2 |

Q4.

| Question number | Answer | Additional guidance | Mark |
|-----------------|---|---|--------------------------|
| CS2 | 33 days is 3 half-lives (1) 2.1(25) × 10 ²² (1) | $\frac{1.7 \times 10^{23}}{2 (\times 2 \times 2)}$ 2.1(25) to any other power of ten scores mp1 only award full marks for correct answer without working. | (2) AO2 |

Q5.

| | Answer | Acceptable answers | Mark |
|------------|--|---|------------|
| (i) | Idea of 2 half-lives (1) 11 400 = 2 × 5700 Idea of halving activity twice (1) 0.55 × 2 × 2 Calculation (1) 2.2 (Bq) | 11 400 / 5700 = 2 2.2 (Bq) for three marks | (3) |

Q6.

| Question number | Answer | Additional guidance | Mark | | | | | | | | |
|-----------------|---|---------------------|--------------|------|---|----------------|----|-----|---------------|--------------------------|------------|
| | <table border="1"> <thead> <tr> <th>mass in g</th> <th>time in days</th> </tr> </thead> <tbody> <tr> <td>1600</td> <td>0</td> </tr> <tr> <td>800 (1)</td> <td>29</td> </tr> <tr> <td>400</td> <td>58 (1)</td> </tr> </tbody> </table> | mass in g | time in days | 1600 | 0 | 800 (1) | 29 | 400 | 58 (1) | numbers in correct boxes | (2) |
| mass in g | time in days | | | | | | | | | | |
| 1600 | 0 | | | | | | | | | | |
| 800 (1) | 29 | | | | | | | | | | |
| 400 | 58 (1) | | | | | | | | | | |

Q7.

| Question number | Answer | Additional guidance | Mark |
|-----------------|---|---|--------------------------|
| | 33 days is 3 half-lives (1) 2.1(25) × 10 ²² (1) | $\frac{1.7 \times 10^{23}}{2 \times 2 \times 2}$ 2.1(25) to any other power of ten scores mp1 only award full marks for correct answer without working. | (2) A02 |

Q8.

| Question Number | Answer | Additional guidance | Mark |
|-----------------|--|--|------------|
| | <p>processing (1)</p> <p>$\frac{125\,000}{1\,000\,000}$</p> <p>OR</p> <p>$\frac{1}{8}$</p> <p>OR</p> <p>3 half-lives or 3×5700</p> <p>evaluation (1)</p> <p>17 100</p> | <p>accept an appropriate attempt using more than one halving</p> <p>17 000</p> <p>award full marks for the correct answer without working</p> | <p>(2)</p> |

Q9.

| Question number | Answer | Additional guidance | Mark |
|-----------------|---|--|------------|
| | <p>recognition of there being 4 half lives involved (1)</p> <p>so fraction of 1/16 involved (1)</p> <p>evaluation (1) 2.4 (kBq)</p> | <p>allow 2 marks for 4.8 (kBq) (used three instead of 4 half lives)</p> <p>allow 1 mark for any other $(1/2)^n$ being involved i.e. for answers that round to 19.3 (kBq), 9.63 (kBq), 1.2(kBq)</p> <p>award full marks for the correct answer without working</p> | (3) |

Q10.

| Question number | Answer | Additional guidance | Mark |
|-----------------|---|--|------------|
| | <p>Determines number of half-lives and rounds (1) $263/87.7 = 3$</p> <p>Determines that 3 half-lives is $1/2 \times 1/2 \times 1/2 = 1/8$ (1)</p> <p>Determines mass of Pu-238 after 3 half-lives (1) $925/8 = 115.625$ (g)</p> <p>Determines average energy released per second (1) $115.625 \times 0.54 = 62.4$ (J)</p> | <p>allow repeated division by 2</p> <p>allow ecf from step 2 for 1 mark (mass of Pu-238 after 1 half-life $925/2 = 462.5$ (g))</p> <p>allow ecf from 1 half-life or from step 3</p> | (4) |